

MICROCOPY RESOLUTION TEST CHART NATIONAL BUREAU OF STANDARDS-1963-A

	SECURITY CUNISHED CHICALTHIS PAGE (When Data Entered,			
REPORT DOCUMENTATION PAGE 'A	READ INSTRUCTIONS BEFORE COMPLETING FORM			
1. REPORT NUMBER 2. GOVT ACCESSION NO	3. RECIPIENT'S CATALOG NUMBER			
AFOSR-TR- 84-1021				
4. TITLE (and Subtitle)	5. TYPE OF REPORT & PERIOD COVERED			
Time-of-flight Spectroscopy of Ionic and	Annual Technical Report			
	June 15, 1983-June 14, 1984			
Metastable Fragments from Dissociating Molecules"	6. PERFORMING ORG. REPORT NUMBER			
AUTHOR(*)	8. CONTRACT OR GRANT NUMBER(#)			
Willis E. Lamb, Jr. L. C. McIntyre, Jr.	FlSf-80-0218			
PERFORMING ORGANIZATION NAME AND ADDRESS	10. PROGRAM ELEMENT, PROJECT, TASK			
Physics Department/Optical Sciences Center	(./// 25			
The University of Arizona	2000			
Tucson, Arizona 85721 1. CONTROLLING OFFICE NAME AND ADDRESS	2001/14			
	12. REPORT DATE			
Air Force Office of Scientific Research	September 25, 1984			
Building 410, Bolling AFB	13. NUMBER OF PAGES			
Washington, DC 20332 14. MONITORING AGENCY NAME & ADDRESS(If different from Controlling Office)	15. SECURITY CLASS. (of this report)			
MONITORING NOCICE HAME & ABBRESS, STREET, STRE	Unclassified			
	Unclassified			
	154. DECLASSIFICATION/DOWNGRADING			
16. DISTRIBUTION STATEMENT (of this Report)	1			
Approved for publi	lc release;			
distribution unli	mited.			
Clstribution unli	mited.			
17. DISTRIBUTION STATEMENT (of the abetract entered in Block 20, if different in Supplementary Notes	Ser Report) STIC SELECTE NOV 2 9 1984			
Clstribution unli	Ser Report) STIC SELECTE NOV 2 9 1984			
17. DISTRIBUTION STATEMENT (of the abetract entered in Block 20, if different in Supplementary Notes	STIC SELECTE NOV 2 9 1984			

84 11 26

26 SECURITY CLASSIFICATION OF THIS PAGE (When D

UNCLASSIFICA

Abstract (continued)

dissociation of molecular ions with a particular inner electron orbital hole. Other experiments described are: (1) an unsuccessful attempt to detect protons from the electron bombardment dissociation of the hydrogen molecular ion; (2) an ongoing attempt to detect, in coincidence, metastable and ionic fragments from dissociation of hydrogen; (3) an experiment on proton fragments from the dissociation of water and hydrogen sulfide.

1	Accession For
ļ	NTIS GREAT
	DTIC T.3
	Unanna
	Just 100 100
(F)/.	By

UNCLASSIFIED

Annual Technical Report

"Time-of-Flight Spectroscopy of

Ionic and Metastable Fragments from Dissociating Molecules"

Principal Investigators

Willis E. Lamb, Jr., Professor of Physics and Optical Sciences

L. C. McIntyre, Jr., Professor of Physics

AFOSR Grant 80-0218

Summary of research June 1983 - June 1984

Research sponsored by the Air Force Office of Scientific Research, Air Force Systems Command, United States Air Force.

Department of Physics and Optical Sciences Center

University of Arizona

Tucson, Arizona 85721

September 25, 1984

AIR FORCE OFFICE OF SCIENTIFIC RESEARCH (AFSC)
NOTICE OF TRANSMITTAL TO DTIC
This technique of the has been for the first approved for the local control of the distribution of the limited.
MATTHEW J. KEWLER

MATTHEW J. King and Chief, Technical Information Division

CONTENTS

		Page
I.	Research Status	1
	A. Introduction	1
	B. Dissociation fragments from the hydrogen halides	1
	C. Dissociation of H ₂ ⁺	2
	D. Coincidence detection of H(2S) and H+	· 3
	from dissociation of H ₂	
	E. Dissociation of H ₂ O and H ₂ S	4
Ħ	Personnel	5

I. Research Status

A. Introduction

This report contains a summary of research in molecular physics conducted under AFOSR grant 80-0218 during the period 15 June 1983 to 14 June 1984. The research was done in the Department of Physics at the University of Arizona by faculty and staff of the Department of Physics and Optical Sciences Center.

During the past several years, we have used time-of-flight methods to measure the velocity distribution of metastable and ionic fragments from molecules dissociated by electron impact. We have concentrated on hydrogen-containing molecules and have observed H(2S) metastable atoms and H⁺, H_2^+ , and H_3^+ ionic fragments. The H(2S) metastable fragments are detected by quenching in an electric field followed by observation of the resulting Lyman- α radiation. The ions are detected in a large-entrance-aperture lens system which focuses ions onto a channel-electron-multiplier. This ion detector can be used in conjunction with a mass filter, 1 developed in this laboratory, which uses a time-dependent potential barrier to obtain a separate velocity spectrum for each ion mass.

Completed experiments include investigation of both the "slow" peak² and the "fast" peak³ in the kinetic energy spectrum of H(2S) fragments from electron bombardment dissociation of H₂. Other published experiments include an investigation of ionic fragments from dissociation of H₂, D₂, and HD,⁴ and a study of H⁺, H₂⁺, and H₃⁺ fragments from methane, ethane, methanol, and ethanol.⁵

Current research will be described in the following sections.

B. Dissociation fragments from the hydrogen halides

This experiment has been completed and a paper is being prepared for submission to the Journal of Chemical Physics. This experiment was the subject of the Ph.D. dissertation of Bruce Kittams who obtained his degree in July 1984. An abstract for the paper is given below.

Time-of-flight and kinetic energy distributions were obtained for H⁺ fragments resulting from electron bombardment of HF, HCl, HBr, and HI at electron energies between 21.2 and 51.2 eV. Several distinct features were observed in these spectra for each molecule and the corresponding threshold electron bombarding energies were obtained. Some of those features are proposed to originate from dissociation of HX+ which is excited by removing an inner σ shell electron in the corresponding molecule. Results from recent (e,2e) experiments support this proposal.

C. Dissociation of H₂+

Considerable effort was made in the past year in an attempt to measure the kinetic energy spectrum of protons from the electron bombardment dissociation of the molecular ion H_2^+ . Although this attempt was ultimately unsuccessful, we will describe here the motivation and the methods used.

The first and only reference to the measurement of the proton kinetic energy spectra from electron impact on H_2^+ is that of Caudano and Delfosse in a conference presentation in 1969.⁶ Previously, Dunn and Van Zyl⁷ had calculated the proton kinetic energy distribution assuming the validity of the first Born approximation while Zare⁸ and Peek⁹ had provided the formal theory of electron impact dissociation of H_2^+ . Some analysis of the Caudano and Delfosse results were given in 1971 by Peek and Green. ¹⁰

Since the early 1970's, the proton kinetic energy spectra from electron impact on H_2^+ has been virtually ignored in both theory and experiment. However, much progress has been made in measuring cross sections for various electron impact processes with H_2^+ , notably the series of papers by Peart and Dolder. Progress has also been made in the field of photodissociation of H_2^+ (see, for example, Van Asselt, Maas, and Los¹²).

A schematic diagram of the experiment is shown in Fig. 1. H_2^+ was produced by electron impact ionization of H_2 using a 0.1 usec pulse from an electron gun capable of delivering a direct current of over 10 mA at 100 V. This gun is called the source gun. The unwanted protons from the initial impact of neutral H_2 were blocked from the detector by a conical electrode to which a 1 usec "cleaning" voltage pulse was applied. Since the H_2^+ produced by the source gun has a thermal distribution of velocities, in principle, there is enough H_2^+ in the interaction region so that following the cleaning pulse, a second electron gun called the "probe" can be pulsed through the interaction region to dissociate the H_2^+ . Protons from the dissociation of H_2^+ then drift through a 12-cm field-free time-of-flight region into a 0.27 steradian collection cone between 17 and 24 degrees in the forward direction. Protons from the dissociation of H_2^+ are then focussed into a Channeltron detector by a cylindrical mirror lens. The source gun must be operated at energies well below 18 eV to avoid dissociation of residual neutral H_2^+ .

After numerous attempts, this experiment has been abandoned because of extremely low counting rates and background problems.

D. Coincidence detection of H(2S) and H+ from dissociation of H2

We are presently attempting to detect, in coincidence, H:(2S) and H+ fragments which move in opposite directions from an interaction region where an election beam crosses an H_2 gas target. There are at least two known excited, dissociating states of H_2 + that come apart yielding these two fragments. We hope to eventually be able to measure the kinetic energy distribution of these fragments using our usual time-of-flight method.

We have constructed an experimental arrangement which has oppositely directed flight paths extending from a central interaction region. Preliminary tests are underway.

E. Dissociation of H₂O and H₂S

We are well underway on a study of proton kinetic energy distributions from dissociation of H_2O and H_2S at electron energies from 20 to 50 eV. We are using our computer-controlled electron gun to measure threshold bombarding energies for observed features in the kinetic energy spectrum. Examples of preliminary data are shown in Fig. 2.

References

- M. D. Burrows, S. R. Ryan, W. E. Lamb, Jr., and L. C. McIntyre, Jr., Rev. Sci. Instrum. 50, 1278 (1979).
- 2. S. R. Ryan, J. J. Spezeski, O. F. Kalman, W. E. Lamb, Jr., L. C. McIntyre, Jr., and W. H. Wing, Phys. Rev. A 19, 2192 (1979).
- J. J. Spezeski, O. F. Kalman, and L. C. McIntyre, Jr., Phys. Rev. A <u>22</u>, 1906 (1980).
- M. D. Burrows, L. C. McIntyre, Jr., S. R. Ryan, and W. E. Lamb, Jr., Phys. Rev. A 21, 1841 (1980).
- 5. M. D. Burrows, S. R. Ryan, W. E. Lamb, Jr., and L. C. McIntyre, Jr., J. Chem. Phys. 71 4931 (1979).
- 6. R. Caudano and J.M. Delfosse, Sixth ICPEAC, Abstracts (1969).
- 7. Dunn and Van Zyl, Phys. Rev. <u>154</u>, 40 (1967).
- 8. R. N. Zare, J. Chem. Phys. <u>47</u>, 204 (1967).
- 9. J. M. Peek, Phys. Rev. 154, 52 (1967).
- J. M. Peek and T. A. Green, Seventh ICPEAC, Abstracts (1971).
- 11. B. Peart and K. T. Dolder, J. Phys. B 4, 1496 (1971); 5, 1554 (1972); 5, 860 (1972); 6, 2409 (1973); 7, 236 (1974).
- 12. N. P. F. B. Van Asselt, J. G. Maas, J. Los, Chem. Phys. Lett. 24, 555 (1974).

II. Personnel

We list below personnel associated with this research program:

Principal investigators:

W. E. Lamb, Jr.

B.S. (Chemistry) University of California (Berkeley) 1934
Ph.D. (Physics) University of California (Berkeley) 1938
D.Sc. University of Pennsylvania 1953
M.A. (by decree) Oxford University 1956
M.A. (hon.) Yale University 1961
L.H.D. Yeshiva University 1965

D.Sc. Gustavus Adolphus College 1975

L. C. McIntyre, Jr.B.S. (Physics) Stanford University 1957M.S. (Physics) University of Wisconsin 1961

Ph.D. (Physics) University of Wisconsin 1965

Faculty Associate:

K. C. Hsieh B.A. (Physics) Wabash College 1963 Ph.D. (Physics) University of Chicago 1969

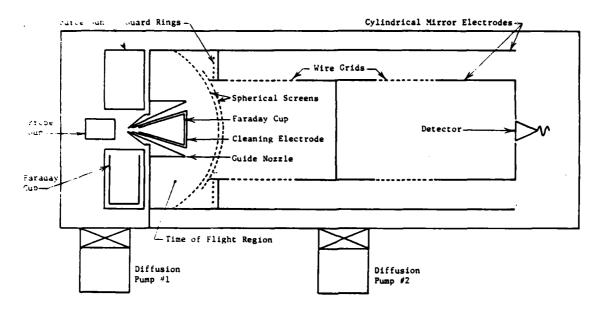
Research Assistants:

Richard Cordaro

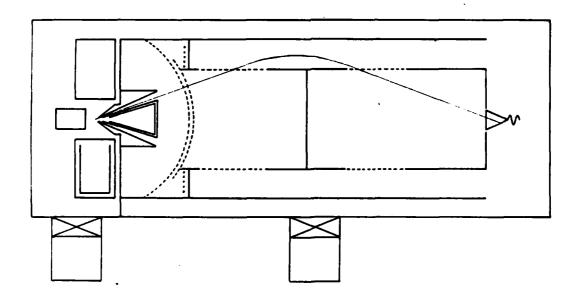
B.A. Millersville St. College, Penn. 1973 M.S. (Physics) Lehigh University 1978

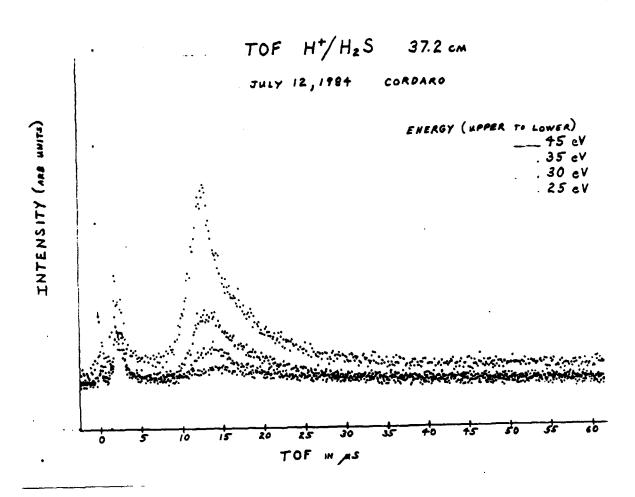
Bruce Kittams (left Aug. 1984)
B.S. Oregon State University 1971
M.S. (Physics) University of Arizona 1981

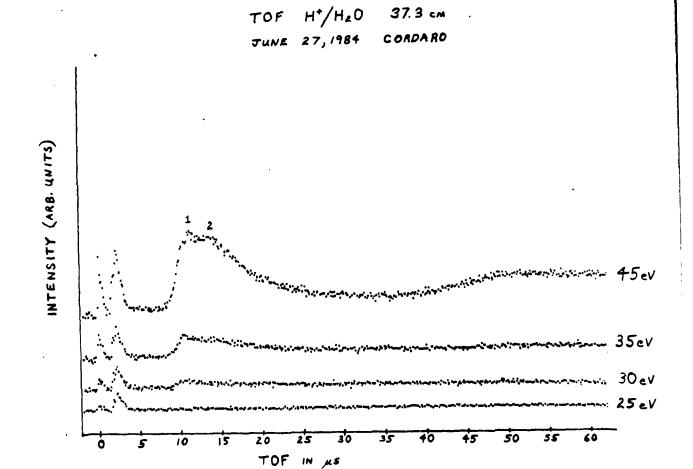
Hyuck Cho
B.S. Sogang University, 1975
M.S. Sogang University, 1977



Sample Ion Trajectory (red)







FILM:

12-84

DTIC